

# Recycling metals from urban mines: what ILs can do and what they cannot do.

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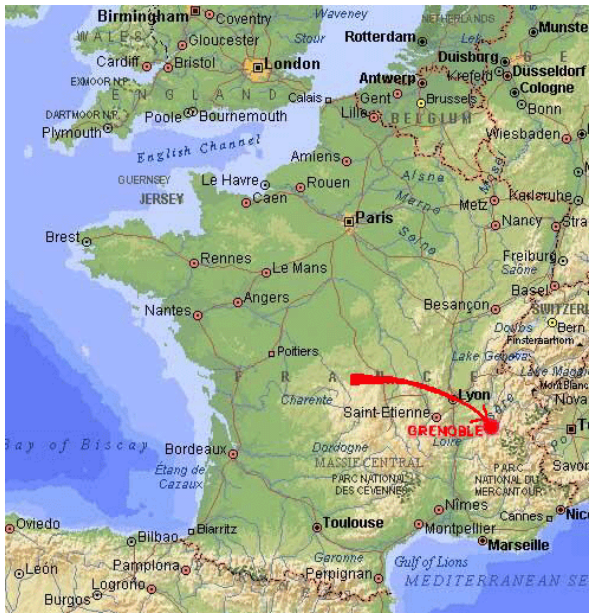
# Outline

- Science in Grenoble and at Lepmi/recycling group
- Urban mines and metal wastes: new challenges
- ILs : what did they achieve ? A short review
- ILs: what can't they perform ?
- Conclusion and perspective

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# Grenoble, France : science and technology at the mountains







European photon & neutron science N-W campus



Institute for geosciences and environment



East campus



STMicroelectronics



# Research at Lepmi, recycling group

- Fundamental aspects:

- Second lifes and aging of batteries
- New liquid/liquid extraction systems (ILs, ABS, DES ...)

(extraction mechanisms, mass transfer at the interface...)

- Current applied topics:

- Recovery of metals (Co, Li etc.) from drone batteries
- Recovery of Ag and cardboard from printed cardboard circuits
- Valorisation of fine incineration ashes
- H<sub>2</sub> bioproduction



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# Every day life with (stupid?) objects

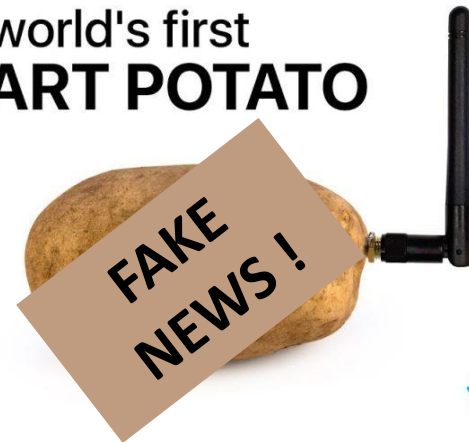




Yes, in my opinion, some are very stupid !



The world's first  
**SMART POTATO**

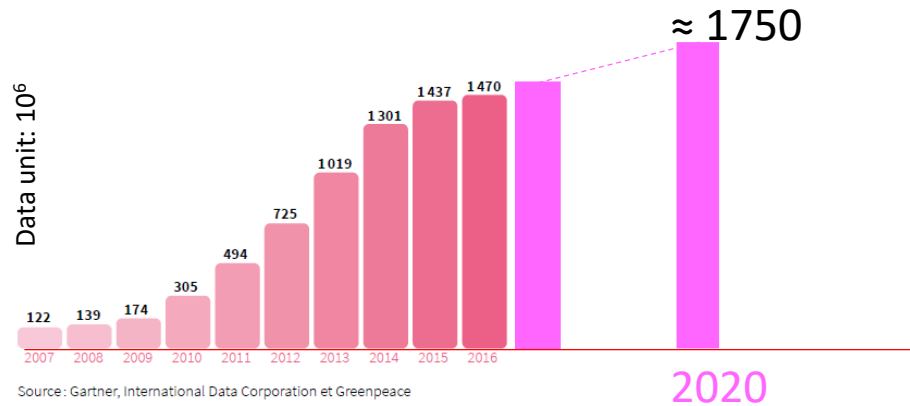


# Consequence: Metal consumption

IA																				VIII															
1																			2																
H																			He																
3		4																5		6		7		8		9		10							
Li	Be															B	C	N	O	F	Ne														
11		12																13		14		15		16		17		18							
Na	Mg															Al	Si	P	S	Cl	Ar														
19		20		21		22		23		24		25		26		27		28		29		30		31		32		33		34		35		36	
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr																		
37		38		39		40		41		42		43		44		45		46		47		48		49		50		51		52		53		54	
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe																		
55		56		57		72		73		74		75		76		77		78		79		80		81		82		83		84		85		86	
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn																		
87		88		89		104		105		106		107		108		109		110		111		112		113		114		115		116		117		118	
Fr	Ra	Ac	Rf	Db	Sg	Ph	Hs	Mt	Ds	Rg																									

Lanthanides													
58	59	60	61	62	63	64	65	66	67	68	69	70	71
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
Terres rares													
90	91	92	93	94	95	96	97	98	99	100	101	102	103
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr



≈ 13.4 BILLIONS smartphones sold worldwide up to 2020.

≈ Two per each of the 7.8 billions humans of 2020 ?

# What is the fate of these discarded objects ?

Discarded objects become WEEE: **W**aste **E**lectronic and **E**lectrical **E**quipment

- In France
  - 835 M EEE items sold in France in 2017, corresponding to 1.88 Mt of EEE
  - 0.69 Mt of household waste, + 0.06 Mt of professional WEE collected in 2017
  - Concerned recycling is difficult, most often unprofitable, and dangerous
- World wide
  - In 2009, 50 Mt of computer wastes discarded (EU: 6 Mt; USA: 21 Mt)
  - Uncontrolled landfill
  - Off shoring of low-end recycling
  - According to WHO, over 7 M humans died in 2012 owing to air pollution
- European/International regulations
  - Basel convention (1997)
  - Restriction of Hazardous Substances #1 and #2 (2003/2006; 2011/2013)



# Specific challenges brought by these new objects

## Many reasons to make recycling difficult and unprofitable

Many components (ceramics, polymers, cardboard, cotton, metals, oxides...)

Metal mixtures unknown in traditional ores

Composition is varying at a very high rate due to innovation

Flows of discarded objects are not constant in time and space



## Current state of metal recycling

We know how to do it !

And we do recycle (many) things !

Low ratio of recycling (at best :30% for PGM)

Polluting ways are implemented

Low rate of recycled materials in new objects

Today's processes should become greener,  
avoiding Volatile Organic Compounds  
and limiting pollution

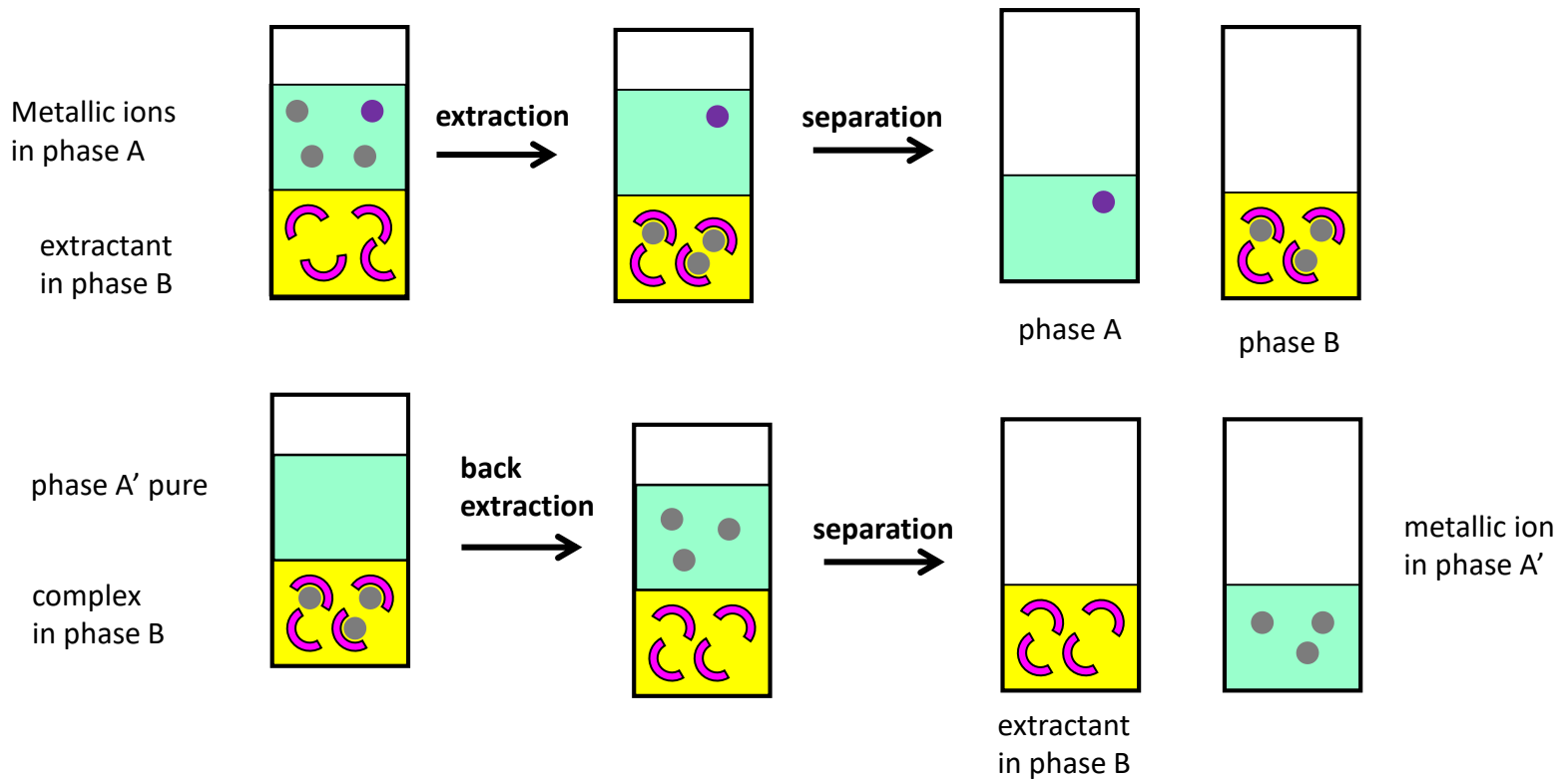


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In this presentation, ILs are salts with a melting temperature below or equal to 100°C.

# Principle of traditional liquid/liquid extraction



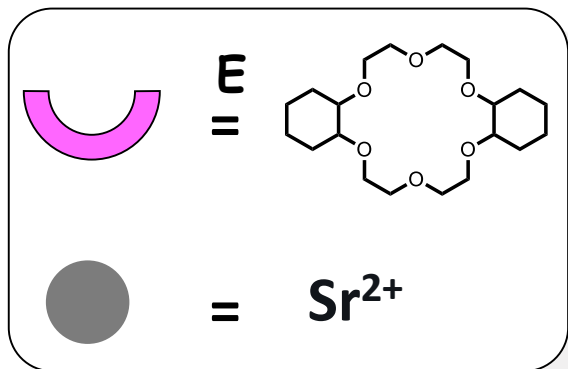
**distribution ratio D:**  $D_M = [M]_{org}/[M]_{aq}$

- $D < 0.1$  no significant extraction
- $0.1 < D < 1$  weak extraction
- $D > 1$  significant extraction

**Phase 1 // Phase 2**  
**Phase 1 can be :**  
**M(o.n)/acid/solvent**  
**Phase 2 can be :**  
**Extractant/solvent**

# 1999: a decisive fundamental work

Sr(II)/pH=4.1/H<sub>2</sub>O//DCH18C6/solvent



solvent	D, no E	D, with E
C <sub>6</sub> H <sub>5</sub> CH <sub>3</sub>	0	0.76
CHCl <sub>3</sub>	0	0.77

**ILs appear as miraculous solvents.  
This is the main basis of recycling studies**

# Magnets: what is inside ?

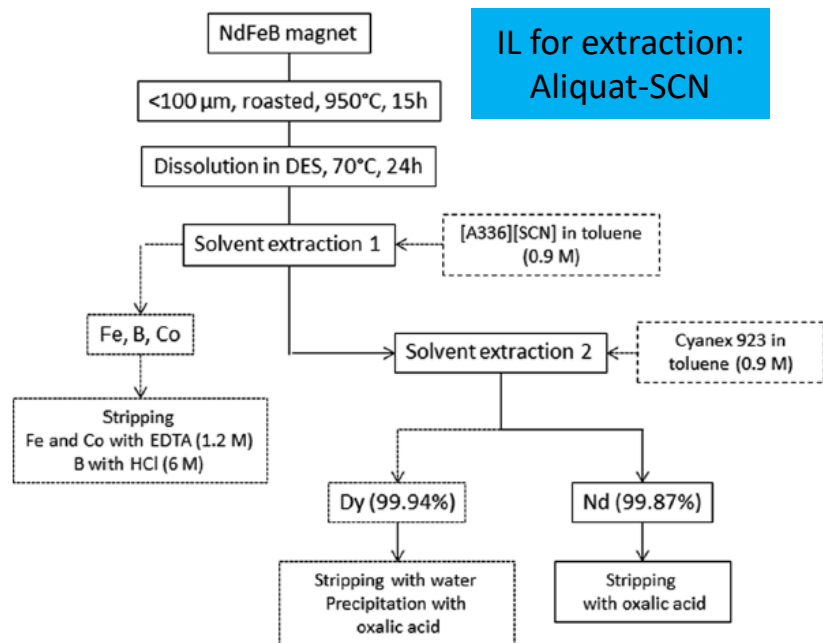
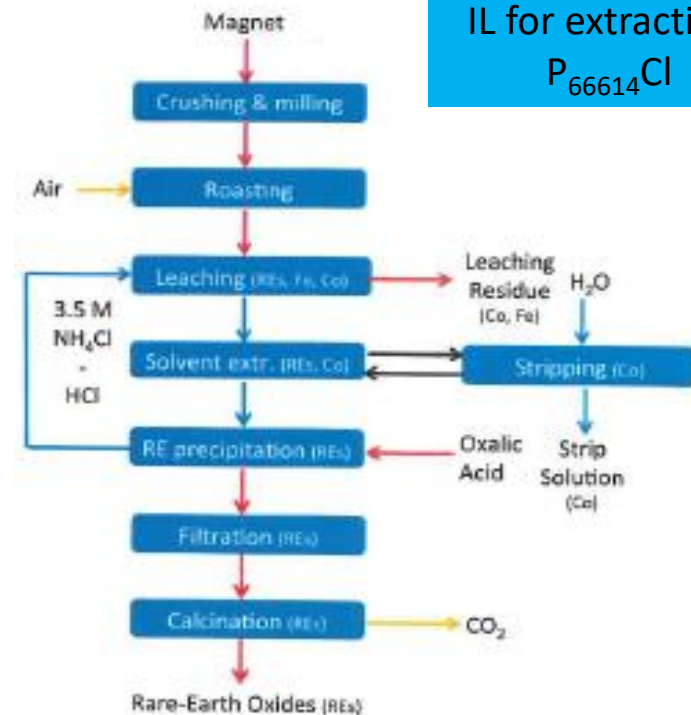
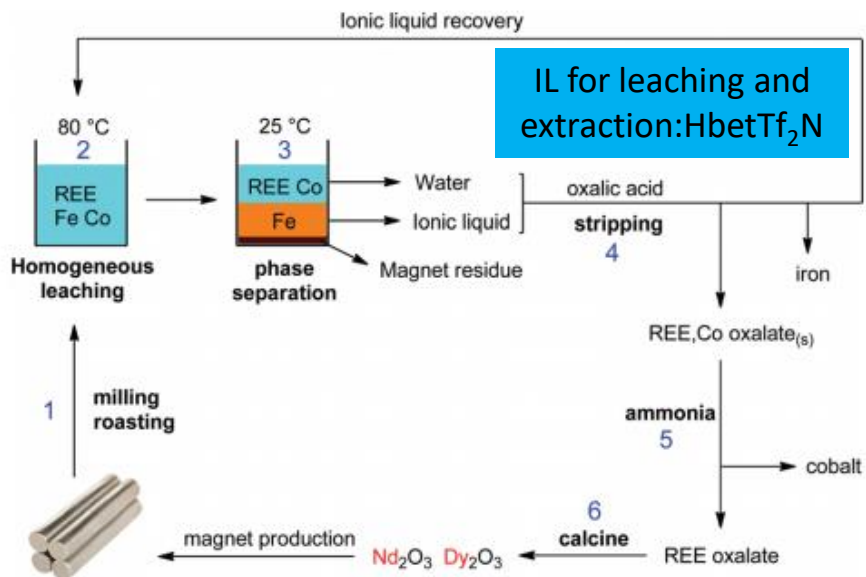
NdFeB and SmCo magnets are sharing most of the production.

Table 1 Composition of the NdFeB magnet in wt%.

Fe	58.16	Pr	0.34
Nd	25.95	C	0.07
Co	4.22	Si	0.06
Dy	4.21	Mn	0.05
B	1.00	Cu	0.04
Nb	0.83	Ni	0.02
O	0.41	N	0.02
Al	0.34	Total	95.72



# Magnets: recycling using ILs



Leaching can be performed in Cl<sup>-</sup> or NO<sub>3</sub><sup>-</sup> medium.

Dupont et al. *Green Chem.*, 17(2015)2150.  
 Riano et al., *RSC Adv.*, 7 (2017)32100.  
 Riano et al., *Green Chem* 17(2015)2931  
 And other papers from the Binnemans' group

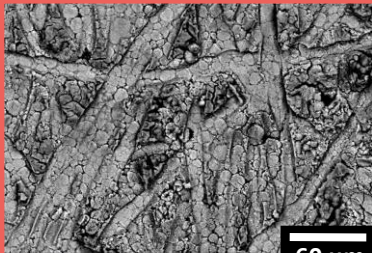
# NiMH batteries: what is inside ?

**SUPPO**



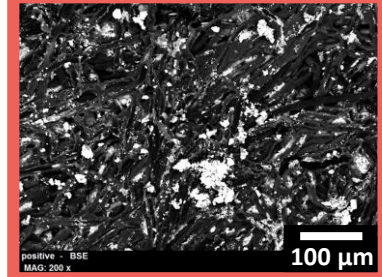
(at. %)  
**Ni: 50.2**  
**Co: 3.4**  
**Mn: 1.2**  
**K: 40.8**  
**REE: -**

**Positive electrode**



**Negative electrode**

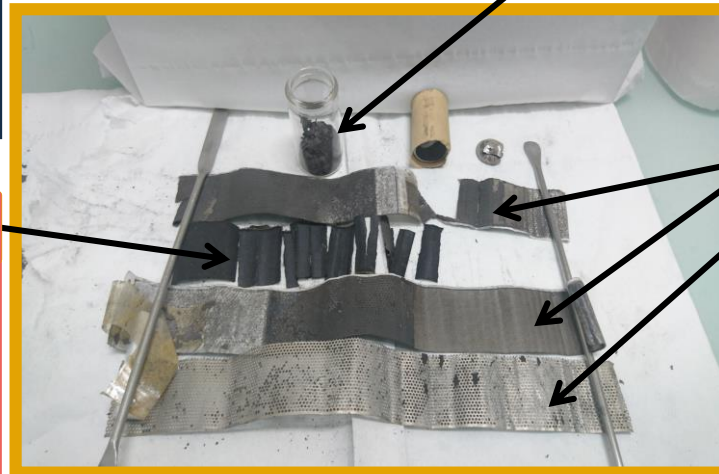
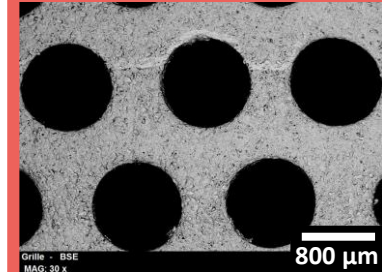
(at. %)  
**Ni: 30.1**  
**Co: 7.6**  
**Mn: 3.7**  
**K: 53.4**  
**REE: 5.2**



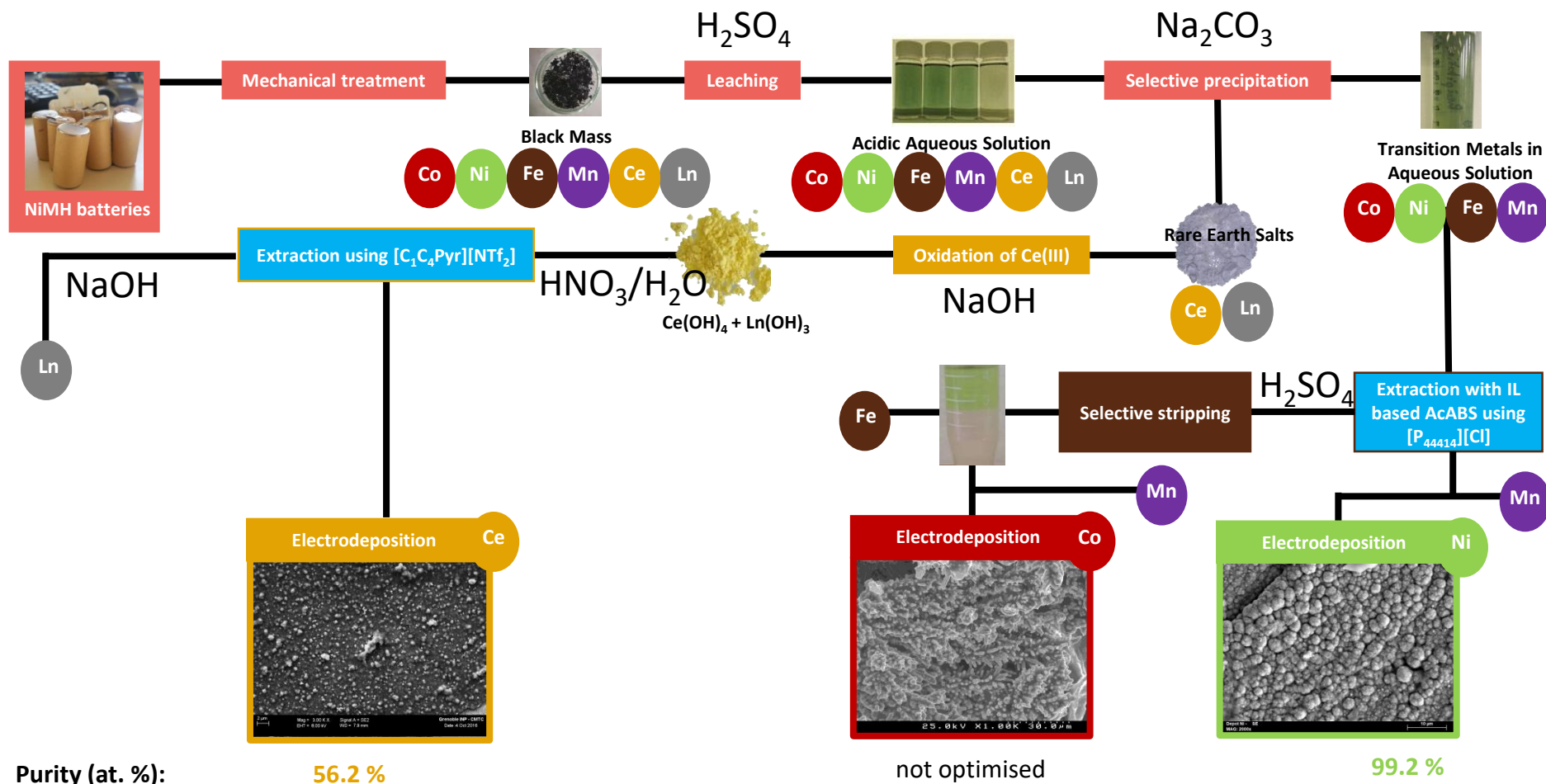
**Separators**

**Metallic grid**

(at. %)  
**Fe: 27.3**  
**Ni: 72.7**



# NiMH batteries: recycling using ILs



Purity (at. %):

56.2 %

not optimised

99.2 %

Directly from  
 $C_1C_4pyrNTf_2$

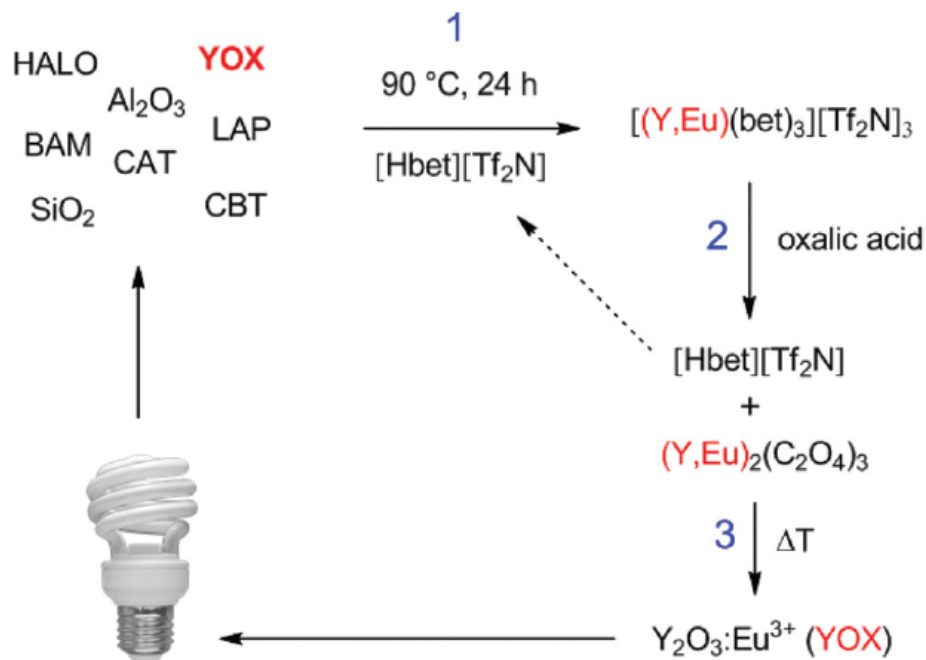
Gras et al., *Sep. Pur. Technol.* 178(2017)169  
Gras et al., *Angew. Chem. Int. Ed.*, 57(2018)1563

# Phosphor lamps

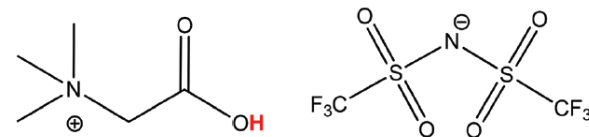
**Table 1** Overview of the lamp phosphors considered in this work and the approximate content found in lamp phosphor waste<sup>3</sup>

Name	Formula	Waste fraction <sup>a</sup> (wt%)	Value
HALO	(Sr,Ca) <sub>10</sub> (PO <sub>4</sub> ) <sub>6</sub> (Cl,F) <sub>2</sub> :Sb <sup>3+</sup> ,Mn <sup>2+</sup>	40–50	Low
YOX	Y <sub>2</sub> O <sub>3</sub> :Eu <sup>3+</sup>	20	High
BAM	BaMgAl <sub>10</sub> O <sub>17</sub> :Eu <sup>2+</sup>	5	Low
LAP	LaPO <sub>4</sub> :Ce <sup>3+</sup> ,Tb <sup>3+</sup>	6–7	High

<sup>a</sup> Approximate fraction found in lamp phosphor waste; the remaining consists of SiO<sub>2</sub> (as fine glass particles), Al<sub>2</sub>O<sub>3</sub>, and small quantities of phosphors like CAT which behaves similarly to BAM, and CBT.



**Fig. 13** Overview of the proposed recycling process for lamp phosphor waste, based on the selective dissolution and revalorization of YOX with the protonated functionalized ionic liquid [Hbet][Tf<sub>2</sub>N]. Step 1: selective dissolution of YOX, step 2: stripping with pure oxalic acid (regenerates the ionic liquid), step 3: direct resynthesis of YOX.



**Fig. 1** Structure of the ionic liquid betainium bis(trifluoromethylsulfonyl)imide, [Hbet][Tf<sub>2</sub>N]. The acidic proton of the betaine group is highlighted in red.

Dupont et al., *Green Chem.*, 17(2015)856



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# Listing ILs disappointments

## **ILs are too costly, even if you recycle them very efficiently**

The cost of the IL amount sequestered is unacceptable for companies

## **ILs are still too toxic and have too much environmental impacts**

Life Cycle Analysis is mandatory and is mainly not in favor of ILs

## **ILs are *not* « designer solvents »**

A provocative opinion, I know...

## **ILs, once diluted, behave as independent ions**

Loss of compounds, synthesis of undesired other ILs etc.

## **New IL usages do not solve the above problems**

ABS, DES etc.

## **ILs do not brew you coffee... they are not the panacea for any recycling problem**

They are part of the solutions, together with acids, bases, heat etc.

In fact, *recycling* is a second best option.  
Better usages, eco-design, second lifes...  
it all depends on you/us !

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# What shall we do next ?

- **Fundamental science**

- Mandatory for understanding (plus : it is fun !)
  - Extraction mechanisms (is unification possible?)
  - Thermodynamics and kinetics of mass transfer

- **Applied studies**

- Very useful for the planet (i.e. : feel good activity)
  - Butts and paper wastes (Jeong et al. ACS Sust. Chem. Eng. 6(2018)4510)
  - All other liquid wastes (industries, wastewaters etc.)
  - Food packaging (Kaiser et al., Recycling, 3(2018)1)

**taking a step back  
from these issues  
is vital**

Pickling baths: Sinoimeri et al., PCCP, 22(2020)23226 + patent.

# Matching green chemistry, costs and science ?

## 1. general thoughts

- What does green chemistry means ? (make our planet great again, yes we can etc...)
  - 'Better', 'greener' are words, not scales or limits
- « Money money makes the world happy »
  - Still true in 2021 but should it stay that way ? (climate change)
  - Regulations at rescue ? (REACH and others ?)
  - Public opinion ? (we/you/I have a role *and* a responsibility)

Doing science may guide  
political choices



# Matching green chemistry, costs and science ?

## 2. one applied limited suggestion

- ILs studied in lab are not cheap, not green, in general
  - Functionalisation often (always ?) worsens the problem
  - Looking for green non toxic ILs is a real difficult job
- **Some industries already did the job !**
  - Plenty of fully non toxic chemicals in catalogs
  - Compounds are already approved, tested, secured etc.
  - Compounds are cheap and available in ton scales
  - Who cares a compound is an IL or is not, if it makes the job ?

Convincing industries to share  
knowledge and extend competences  
is a difficult frustrating experience  
but it is worth it !

# Institutions, fundings and supports



Co-leading of an Interdisciplinary Research program #1 :  
eco-design, second lifes, recycling



European project : Novel methods for enhanced recovery of metals  
and minerals from fine incineration ash

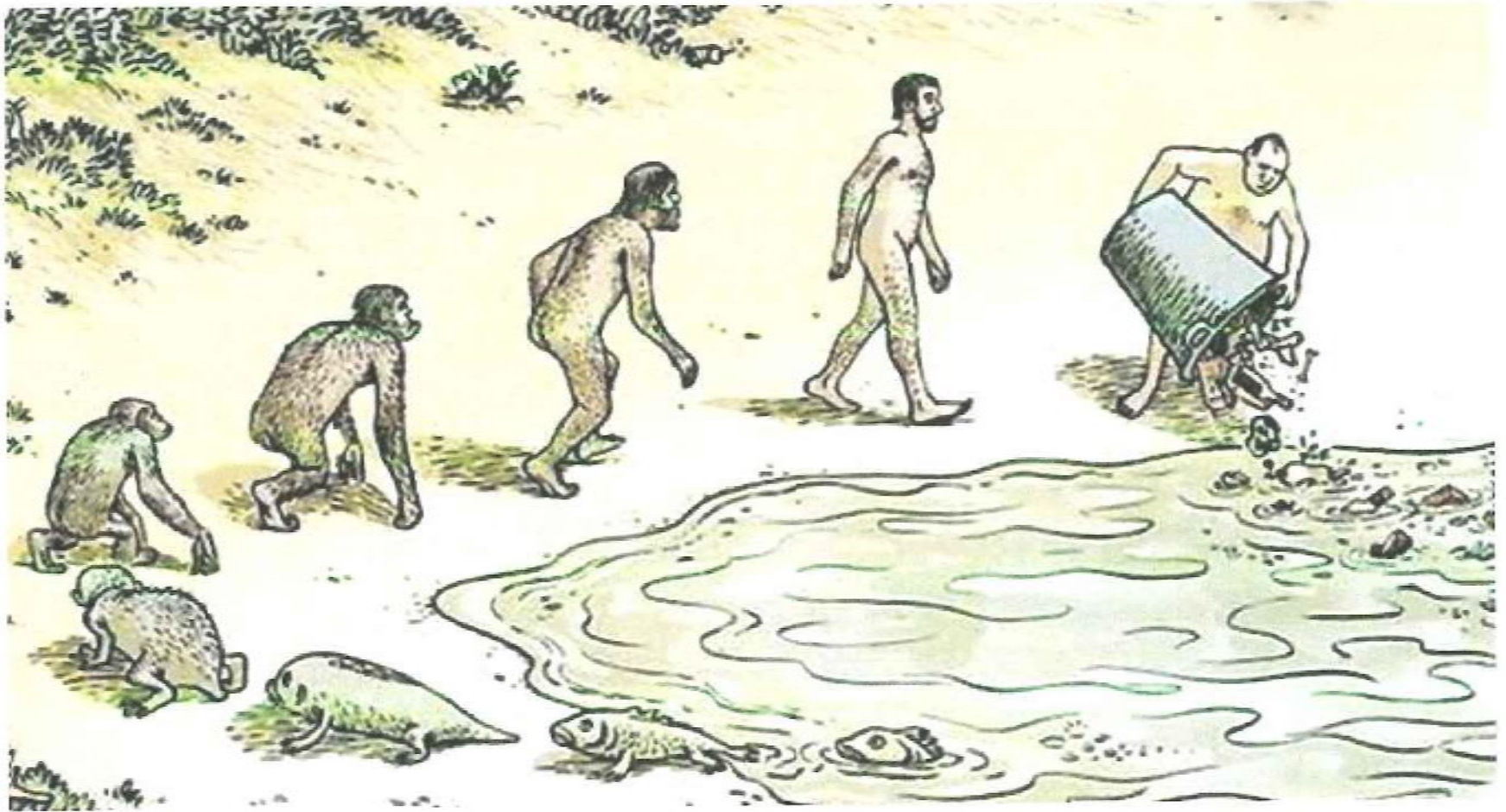


Steel company. Recycling of valuables from pickling used baths



Multidisciplinary initiatives: ENiDES (Extraction of Ni with DES,  
using plants for polluted soils)

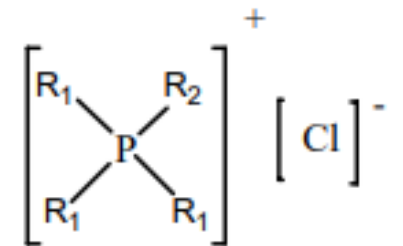
# Thanks for your attention !



# Aqueous biphasic systems (ABS): a chemical (r)evolution

ABS are liquid mixtures being either mono or biphasic depending on chemical composition and temperature

- 1958: Uppsala, Sweden: H<sub>2</sub>O/polymer/inorganic salt
  - Used for separation of proteins
  - Poor efficiency
- 2003: Tuscaloosa, USA: H<sub>2</sub>O/ionic liquid/inorganic salt
  - Used for separation of short chain alcohols
  - Moderate efficiency
- 2018: Grenoble, France: H<sub>2</sub>O/ionic liquid/mineral acid
  - Used for the separation of metallic ions
  - High efficiency

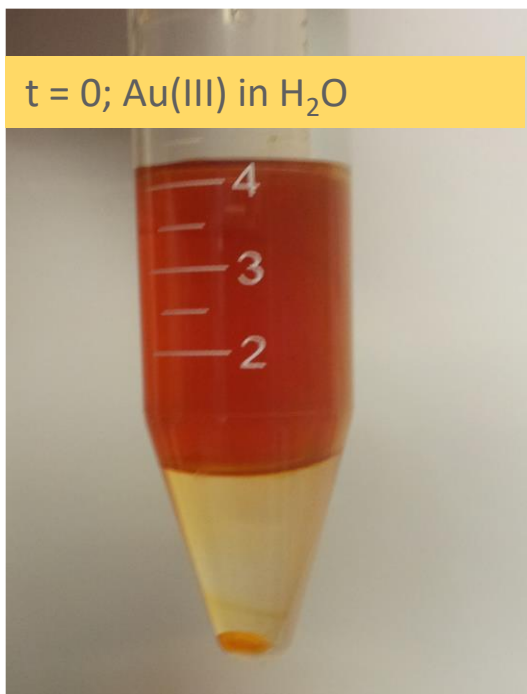


R<sub>1</sub>: butyl C<sub>4</sub>  
R<sub>2</sub>: tetradecyl C<sub>14</sub>

C<sub>1</sub>C<sub>4</sub>imCl and P<sub>44414</sub>Cl are  
hydrophilic ionic liquids

# ILs do not need extractants

Dai et al. first demonstrated that « simple » ILs can extract metal ions without the help of extractant.



N. Papaiconomou et al., *Green Chem.* 14(2012)2050  
Papaiconomou et al., *RSC Advances*, 4(2014)48260-48266



## Extraction mechanisms can be very different from those in molecular solvents

In molecular solvents, extracted species are neutral

In ILs, they can be positive, negative or neutral.

Ion exchange means loss of IL ions and aqueous phase pollution together with water transfer to the other phase

### U(VI)/HNO<sub>3</sub>/H<sub>2</sub>O//TBP/solvent

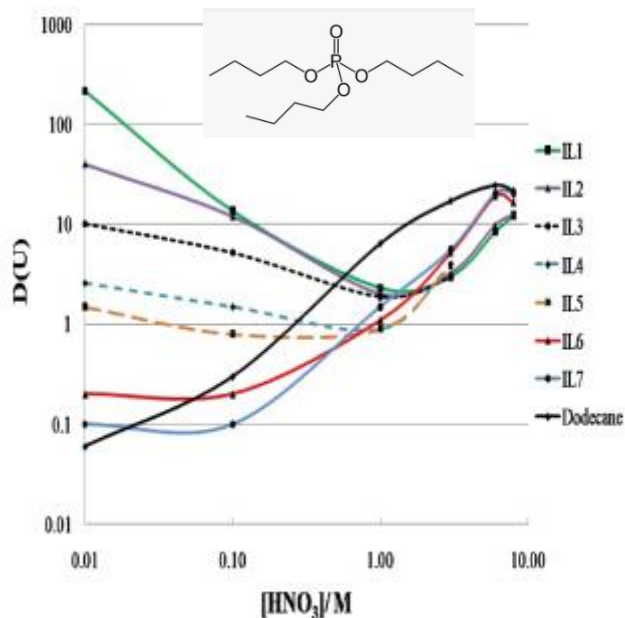
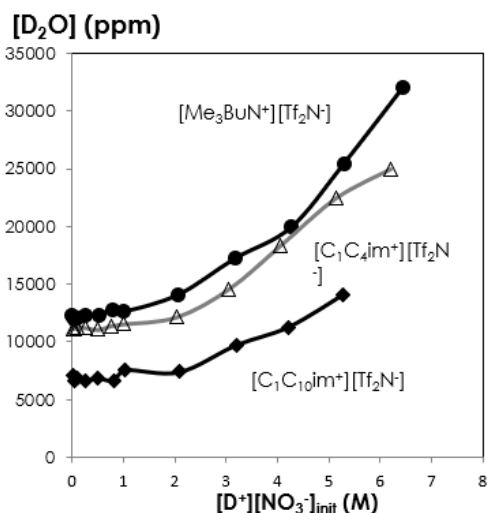
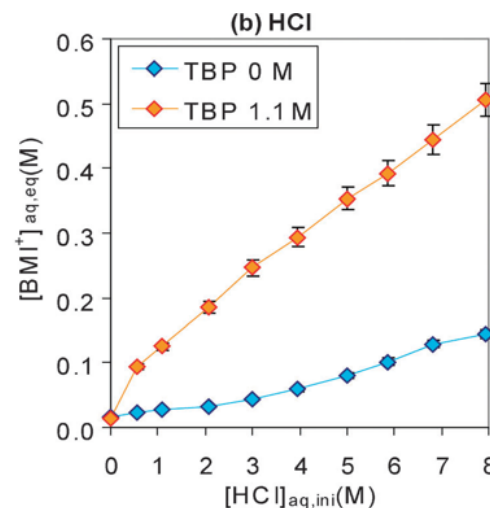


Fig. 2 Dependency of  $D(U)$ , on  $[HNO_3]$  at constant TBP (1.2 M) concentration in ionic liquids and dodecane.



Mazan et al., RSC Advances, 4(2014)13371;



Bell et al. Dalton Trans. 40(2011)10125.

Extraction mechanisms, review papers by : Dietz Sep Sci. Technol., 41(2006)2047; Billard et al., Anal. Bioanal. Chem., 400(2011)1555; Stojanovic et al., Sep. Sci. Technol., 47(2012)189; Kolarik, Solv. Ext. Ion Exch. 31(2013)24;  
Mutual solubilities: Rickert et al., Talanta, 72(2007)315; Gaillard et al., PCCP, 14(2012)5187-5199; Fagnant et al., Inorg. Chem. 52(2013)549.